

Chapter 17 Reinforcement Acids Bases Answers



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chapter 23 section 1 and 2- science. Quiz: 5/23/13: LAST DAY OF SCHOOL!!! STUDY. PLAY. Terms in this set (...) acid. sour taste. ... solutions containing ions that react with additional acids or bases to minimize their effects of pH. Acids and Bases 69 terms. baegae000. Chapter 22-Acids, ... Chapter 5 physical science vocab 18 terms. aledy12 ...

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A conjugate base is the species that results when an acid donates a hydrogen ion. The acid HX donates its hydrogen ion and becomes the conjugate base X⁻. In the reaction shown above, the hydronium ion (H₃O⁺) is the conjugate acid of the base H₂O. The X⁻ ion is the conjugate base of the acid HX.

Chapter 18: Acids and Bases - FCPS

is the base) 9.2 Bronsted-Lowry Acids and Bases. Bronsted-Lowry acid-base reactions involve a transfer of a proton (H⁺) from an acid to a base. o Bronsted - Lowry Acid: proton donor (loses H⁺) o Bronsted - +Lowry Base: proton acceptor (gains H) □ This definition allows for a broader range of bases to be included.

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