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-The coefficients in a balanced chemical equation specify the relative amounts in moles of each of the substances involved in the reaction-involves the mass relationships between reactants and products in a chemical reaction (based on chemical equations and the law of conservation of mass).

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A hydrogen bond is the attraction between the lone pair of an electronegative atom and a hydrogen atom that is bonded to either nitrogen, oxygen, or fluorine. The hydrogen bond is often described as a strong electrostatic dipole-dipole interaction. However, it also has some features of covalent bonding: it is directional, stronger than a van der Waals force interaction, produces interatomic ...

Intermolecular force - Wikipedia

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